

CIS-OXYPALLADATION COMPLEXES OF A PINENE DERIVATIVE

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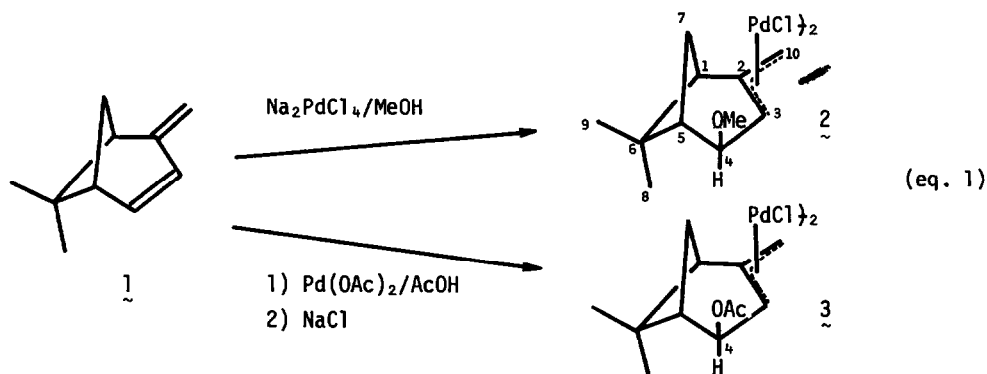
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Abstract. 4-Methoxy- and 4-acetoxy(3,2,10- η -pinene)palladium complexes prepared from dehydro- β -pinene with Na_2PdCl_4 in MeOH and $\text{Pd}(\text{OAc})_2$ in AcOH, respectively, represent the first isolated *cis*-oxypalladation adducts.

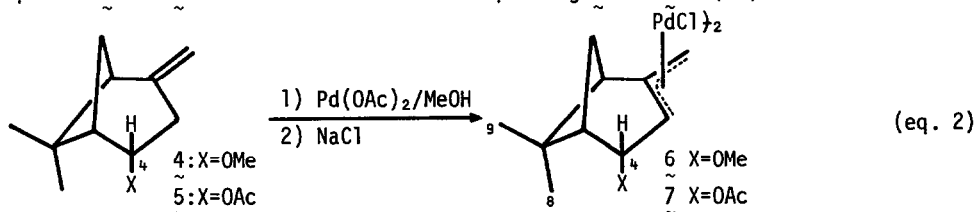
Palladium(II)-induced nucleophilic addition to olefins has been extensively studied from both synthetic and mechanistic aspects.¹⁾ In particular, the stereochemical mode of the addition has attracted much attention. It appears to be one of general agreements that oxygen nucleophiles such as acetate, alcohol, and water attack on π -olefin and σ -alkylpalladium complexes in a *trans*-fashion.²⁾ Recently, a new observation on this regard has been put forward in which acetate can add *cis* for π -allylpalladium complexes.³⁾ This result prompts us to report our own work on the first isolation of *cis*-oxypalladation complexes derived from the attack of alcohol and acetate on a pinene derivative co-ordinated to palladium.

During the course of our investigation on the Pd(II)-catalyzed asymmetric cyclization of 2-allylphenols,⁴⁾ we have prepared a variety of η^3 -pinenepalladium complexes. When dehydro- β -pinene (**1**) was allowed to react with an equimolar amount of Na_2PdCl_4 in MeOH at room temperature for 1 hr, the π -allyl complex **2** was obtained as a single product in 53% isolated yield.⁵⁾ Treatment of **1** with $\text{Pd}(\text{OAc})_2$ in AcOH followed by addition of NaCl gave the complex **3** in 47% yield. The reaction of **1** with $\text{Pd}(\text{OAc})_2$ in MeOH under the same conditions resulted in the formation of both complexes **2** and **3** in 12 and 14% yields, respectively.

The following results allow us to assign that **2** and **3** are the *cis*-oxypalladation complexes as depicted in eq. 1. (i) π -Allylpalladium complexes **6** and **7**, which have the *trans*-configuration



between Pd and the C-4 substituents (OMe and OAc), were independently prepared from the β -pinenes 4 and 5 of known stereochemistry,⁶⁾ as shown in eq. 2. Thus, the complex 6 was obtained in 52% yield from the compound 4 bearing the C-4 substituent (X=OMe) at the same site of the *gem*-dimethyl group, upon treatment with Pd(OAc)₂ in MeOH followed by addition of NaCl. Similarly, 7 was obtained from 5 (X=OAc) in 58% yield. Comparison of the spectral data in the Table shows that the complexes 2 and 3 are different from corresponding 6 and 7. (11) When the relative



chemical shifts for the *gem*-dimethyl group ($\Delta\delta$ 8-Me and 9-Me) of 2, 3, 6, and 7 were compared with those of the corresponding parent α -pinenes, any meaningful shift could not be detected. If the palladium in the complexes is in the same site of the *gem*-dimethyl group, the relative chemical shift must be greatly affected by the proximity between the metal and one of the methyl groups.⁷⁾ Therefore, the configuration between Pd and the C-4 substituent is determined to be *trans* for 6 and 7 and *cis* for 2 and 3.⁸⁾

Table. Representative NMR Shifts of π -Allyl Complexes (100 MHz, CDCl₃, δ)

Complex	8-Me	9-Me	H-10	H-3	H-4	J _{3,4} (Hz)	J _{4,5} (Hz)
<u>2</u>	0.99	1.39	3.03	3.73	3.99	3.52	4.5
<u>3</u>	1.04	1.40	3.08	3.69	4.15	4.88	5.0
<u>6</u>	1.11	1.38	3.08	3.75	3.96	3.65	0
<u>7</u>	1.13	1.39	3.09	3.77	3.89	3.65	0

References and Notes

- 1) For a comprehensive review, see P. M. Henry "Palladium Catalyzed Oxidation of Hydrocarbon", D. Reidel: Holland, 1980; pp 44-223.
- 2) J.-E. Backvall, B. Åkermark, and S. O. Ljunggren, *J. Am. Chem. Soc.*, **101**, 2411 (1979).
- 3) (a) J.-E. Backvall, R. E. Nordberg, E. E. Björkman, and C. Moberg, *J. C. S. Chem. Comm.*, 943 (1980); (b) J.-E. Backvall and R. E. Nordberg, *J. Am. Chem. Soc.*, **103**, 4959 (1981).
- 4) T. Hosokawa, T. Uno, S. Inui, and S.-I. Murahashi, *J. Am. Chem. Soc.*, **103**, 2318 (1981).
- 5) Satisfactory elemental analyses for all the complexes reported herein were obtained. Yields are based on isolated products by preparative TLC (SiO₂, EtOAc n-hexane=1:4).
- 6) G. Ohloff and W. Giersch, *Helv. Chem. Acta*, **60**, 1496 (1977).
- 7) A. D. Buckingham and W. Urland, *Mol. Phys.*, **26**, 1571 (1973). The same argument has been already put forward by B. M. Trost, P. E. Strege, L. Weber, T. J. Fullerton, and J. Dietsche, *J. Am. Chem. Soc.*, **100**, 3407 (1978).
- 8) The complexes 2 and 3 are the primary products in the present reactions, since 6 and 7 are not converted into 2 and 3 under the reaction conditions, respectively.

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